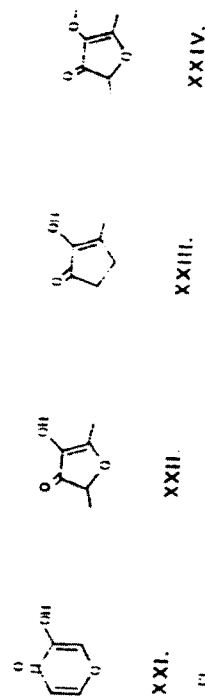


Reaction schemes have been presented for the formation of these heterocyclics (Tress *et al.* 1985a). As is shown in Figure 3.6A, pyruvaldehyde and L-proline form an iminium carboxylate intermediate which is transformed by decarboxylation to a reactive ylide or iminium ion. These two intermediates then may undergo hydration to yield L-pyrroline and acetal or pyrrolidine and pyruvaldehyde. Reduction would yield L-acetonyl pyrrolidine.

The pyrrolizines may arise from proline/dicarbonyl reactions as shown in Fig. 3.6B. This involves aldol type reactions via the iminium ion in intermediate. Further reaction by Michael addition or aldol condensation and dehydration would yield the pyrrolizines.

f. Furanones and Pyranones

The furanones and pyranones are oxygen containing heterocyclic compounds associated with both caramelized and NFII flavors. The odor characters most common to this group of compounds would be caramel-like, sweet, fruity, butterscotch, nutty or burnt. They predominate both in proportion and absolute amount in concentrates of carbohydrates that are subjected to browning reactions (Ohloff and Flament 1979). It is of interest that about 70 furan derivatives have been identified in coffee aroma, 25 in the volatiles of bread and 37 in caramel (Ohloff and Flament 1979).



The caramel character is associated with a planar, contiguous C-alkyl and carbonyl group (alkyl C-C(OH)C=O) in the molecule (Pittet *et al.* 1970). Maltol (XXI) was one of the first compounds in this class to be identified in foods. Ethyl maltol (2-ethyl 3-hydroxy 4(1H) pyranone) also has a caramel odor but is approximately four to six times stronger in flavor strength than maltol. Furanal (4-hydroxy 2,5-dimethyl

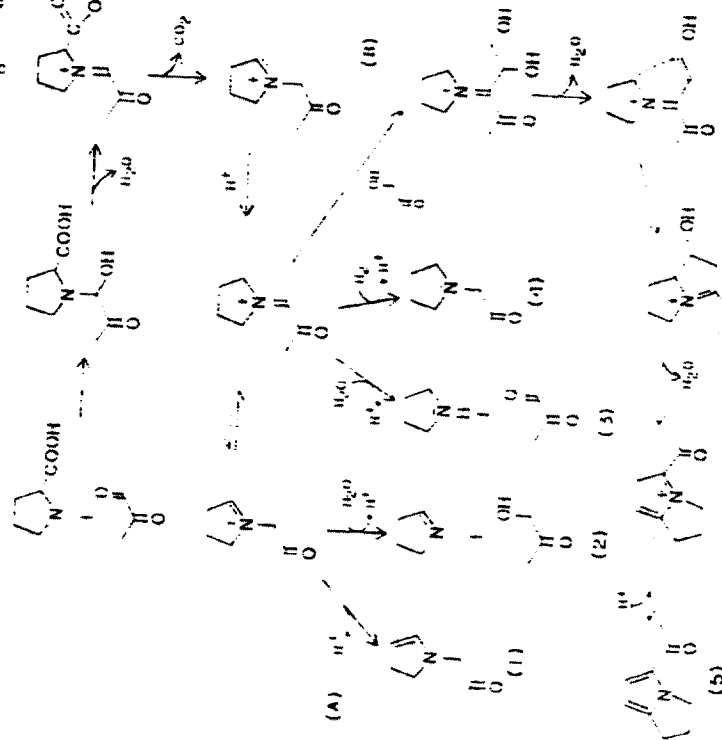


Fig. 3.6. Strecker degradation of proline to yield (1) L-acetonyl pyrrolidine; (2) L-acetyl pyrrolidine; (3) diacetyl pyrrolidine; (4) L-acetonyl pyrrolidine; (5) L-acetyl pyrrolidine.

3(2H)furanone, XXII) is the trademark of a compound used extensively in the flavor industry and, like maltol and ethyl maltol, is a flavor enhancer for sweet products. Furanal itself has a "burnt pineapple" odor.

The five carbon analogs of maltol (XXIII) and furanone possess odor properties similar to their oxygen containing counterparts. Cyclohexene (tradename for compound XXIV) has a very characteristic sweet maple character. The related compound, 3-ethyl 2-hydroxy 2-cyclopenten-1-one, also finds extensive use in the flavor industry for similar applications (caramel, nut, maple and butterscotch flavorings).

Compounds of this class that do not meet the structural requirements for being caramel in character may have quite different sensory prop-

3. Changes in Food Flavor Due to Processing

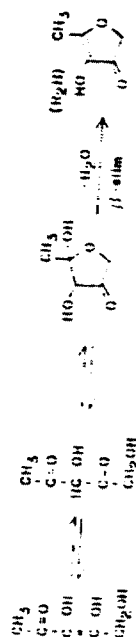
For example, 4-methoxy 2,5-dimethyl 3(2H) furanone (XXIV) has an odor similar to sherry. The *n*-butyl ether version of this furanone has a jasmine-like odor.

This group of compounds does not contain nitrogen. Therefore, the mechanism of formation generally involves the cyclization of nonnitrogen containing browning intermediates (Fig. 3-7). These intermediates may be products of major browning pathways involving sugar dehydration on the Strecker degradation (Reynolds, 1970; Hodge *et al.*, 1972).

g. Sulfur Heterocyclics

Several different types of heterocyclic compounds containing sulfur are produced via NEB. These include thiophenes, dithiols, dithianes, dithians, trithianes, trithianes, tetrahydrothiophenes, thiazoles, thiazolines and

(V)



(II)

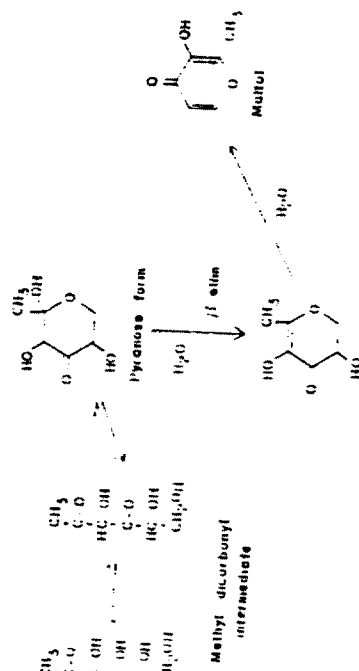


Fig. 17. Mechanisms proposed for the formation of furans and furanones. (A) Formation of a dicyclopentenone. (B) Formation of malloyl.

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thiazolidines. The major heterocyclic sulfur compounds produced via NEB are thiophenes and thiophenes.

Thiophenes and pyrazines have somewhat similar sensory properties. Pittet and Huza (1974) and Ho and Jim (1985) have reported that the alkylthiophenes give green, nutty, roasted, vegetable or meaty notes. Tri-methyl thiophene (XXV) is reported to have a cocoa, nutty character. The 2-isobutyl thiophene (XXVI) is one of the best known as having the strong green odor of tomato leaf. This compound is considered to be important to tomato flavor. The 2,4-dimethyl 5-vinyl thiophene (XXVII) has a nutlike odor. The 2-acetyl thiophene (XXVIII) is characterized as having a nutty, cereal and popcorn flavor (Ohloff and Flament 1979).



XXV.



XXVI.



XXVII.



XXVIII.

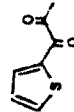
Thiophenes have only recently been identified as food flavors. With the exception of cranberries, thiophenes have only been found in cooked or roasted food products (Ohloff and Flament 1979). The sensory properties of many of the thiophenes have been summarized by Ohloff and Flament (1979). Thiophene (XXIX) has a pungent character while 2,4-dimethyl thiophene is well known for its importance to the flavor of fried onions. The 2-acetyl 3-methyl thiophene (XXX) has a honeylike sensory quality at 0.25 g/100 L concentration but is nutty and starchy at 0.11 g/100 L concentration. Pridine and woody notes are characteristic of compound XXXI. The 5-methyl thiophene-2-carboxaldehyde (XXXII) has a burnt coffee note (Shibamoto 1983).



XXIX.



XXX.



XXXI.



XXXII.

The sensory properties of some of the other sulfur containing heterocyclic compounds are presented in Table 3-2. These compounds, like the thiophenes and thiophenes, typically are formed via the reaction of sulfur containing amino acids with intermediates of the browning reaction. An alternative mechanism involves the initial formation of H₂S with browning intermediates (Sakaguchi and Nishimura 1978). The formation of